

<p><b>2002-646209/70</b>      A96 D21 E19 (A14 A25 A26)      <b>OREA 2000.11.21</b></p> <p>L'OREAL SA      *FR 2816833-A1</p> <p>2000.11.21 2000-015035(+2000FR-015035) (2002.05.24) A61K 7/06, 7/075</p> <p><b>Cosmetic composition comprises polymer dispersion prepared by polymerizing water-soluble monomer in aqueous salt solution containing polyelectrolyte dispersant and viscosity increase inhibitor</b></p> <p><b>C2002-182490</b></p> <p>Addnl. Data: GIROUD F</p>	<p>A(4-F1A, 10-E11, 12-V4A) D(8-B4, 8-B5) E(10-A7, 10-A9B8, 10-A22B, 10-A22D, 10-B2B, 10-C2, 10-C3, 10-D3C, 10-E2D5, 10-E2U, 10-E4K, 32-A4)</p>
<p><b>NOVELTY</b></p> <p>Cosmetic composition comprises a polymer dispersion prepared by polymerizing a water-soluble monomer having at least one double bond in an aqueous salt solution containing a polyelectrolyte dispersant and a selected viscosity increase inhibitor.</p> <p><b>DETAILED DESCRIPTION</b></p> <p>Cosmetic composition comprises a polymer dispersion prepared by polymerizing a water-soluble monomer having at least one double bond in an aqueous salt solution containing a polyelectrolyte dispersant and a viscosity increase inhibitor selected from:</p>	<p>(a) polycarboxylic acids and their salts;</p> <p>(b) polyhydroxylic phenols;</p> <p>(c) cyclic hydroxy carboxylic acids and their salts;</p> <p>(d) gluconic acid and its salts;</p> <p>(e) products obtained by reacting methoxyhydroquinone and/or a cationic (meth)acrylic monomer with a radical source in an oxidizing atmosphere;</p> <p>(f) products obtained by reacting a cationic (meth)acrylic polymer with a radical source in an oxidizing atmosphere; and</p> <p>(g) products obtained by reacting a cationic (meth)acrylic polymer with an oxidizing agent.</p> <p><b>USE</b></p> <p>The composition is useful as a rinse-off or leave-on hair product, especially a shampoo or styling mousse, lotion or gel.</p> <p><b>EXAMPLE</b></p> <p>A typical composition was prepared by polymerizing a mixture of</p> <p>FR 2816833-A+</p>

<p>acryloyloxyethyl dimethyl benzyl ammonium chloride (30 mole%), acryloyloxyethyl trimethyl ammonium chloride (50 mole%) and acrylamide (20 mole%) in a reaction mixture comprising the monomers (25%), pyrogallol (2500 ppm), p-hydroxybenzoic acid (50 ppm), poly(dimethyldiallylammonium chloride) (1%), poly(methacryloyloxyethyl trimethyl ammonium chloride) (1%), ammonium sulfate (19%) and water (to 100%).</p> <p><b>TECHNOLOGY FOCUS</b></p> <p><b>Polymers - Preferred Monomer:</b> The water-soluble monomer is either: (i) a quaternized diallylamine derivative or a quaternary ammonium or acid addition salt of an aminoalkyl (meth)acrylate of (meth)acrylamide in which the alkyl group is ethyl, propyl or hydroxypropyl; (ii) (meth)acrylic acid or 2-acrylamido-2-methylpropane sulfonic acid; or (iii) (meth)acrylamide, N-vinylformamide, N-vinylacetamide or hydroxypropyl (meth)acrylate.</p> <p><b>Preferred Polymer:</b> The polymer comprises either: (i) 20 mole% acrylamide, 30 mole% acryloyloxyethyl dimethyl benzyl ammonium chloride and 50 mole% acryloyloxyethyl trimethyl ammonium chloride; or</p>	<p>(ii) 65 mole% acrylamide and 35 mole% acryloyloxyethyl trimethyl ammonium chloride.</p> <p><b>Preferred Polyelectrolyte:</b> The polyelectrolyte is a product obtained by polymerizing 50-100 mole% of a dimethylaminoethyl (meth)acrylate salt, an N-dimethylaminopropyl (meth)acrylamide salt, a di(meth)allylamine salt, (meth)acryloyloxyethyl trimethyl ammonium chloride or (meth)acrylamidopropyl trimethyl ammonium chloride and 0-50 mole% acrylamide.</p> <p><b>Organic Chemistry - Preferred Viscosity Increase Inhibitor:</b> The viscosity increase inhibitor is preferably: oxalic, adipic, tartaric, malic or phthalic acid; m- or p-hydroxybenzoic acid, salicylic acid, gallic acid or tannic acid; or gluconic acid or a gluconate salt.</p> <p><b>Inorganic Chemistry - Preferred Salt Solution:</b> The salt solution is preferably an ammonium, sodium, magnesium or aluminum sulfate or bisulfate solution.</p> <p>(29pp367DwgNo.0/0)</p>
--	--